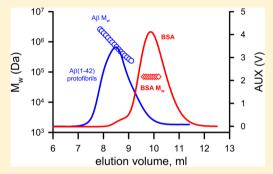


Biophysical Comparison of Soluble Amyloid- $\beta(1-42)$ Protofibrils, Oligomers, and Protofilaments

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ABSTRACT: Some of the pathological hallmarks of the Alzheimer's disease brain are senile plagues composed of insoluble amyloid- β protein $(A\beta)$ fibrils. However, much of the recent emphasis in research has been on soluble $A\beta$ aggregates in response to a growing body of evidence that shows that these species may be more neurotoxic than fibrils. Within this subset of soluble aggregated A β are protofibrils and oligomers. Although each species has been widely investigated separately, few studies have directly compared and contrasted their physical properties. In this work, we examined wellrecognized preparations of $A\beta(1-42)$ oligomers and protofibrils with multiangle (MALS) and dynamic (DLS) light scattering in line with, or following, size-exclusion chromatography (SEC). Multiple SEC-MALS analyses of protofibrils revealed molecular weight (M_w) gradients ranging



from 200 to 2600 kDa. Oligomeric A β species are generally considered to be a smaller and more nascent than protofibrils. However, oligomer M_w values ranged from 225 to 3000 kDa, larger than that for protofibrils. Root-mean-square radius (R_v) values correlated with the $M_{\rm w}$ trends with protofibril $R_{\rm g}$ values ranging from 16 to 35 nm, while oligomers produced one population at 40–43 nm with a more disperse population from 22 to 39 nm. Hydrodynamic radius ($R_{\rm H}$) measurements by DLS and thioflavin T fluorescence measurements indicated that protofibrils and oligomers had commonalities, yet electron microscopy revealed morphological differences between the two. SEC-purified A β (1–42) monomer at lower concentrations was slower to nucleate but formed protofibrils (1500 kDa) or soluble protofilaments (3000 kDa) depending on the buffer type. The findings from these studies shed new light on the similarities and differences between distinct soluble aggregated A β species.

lzheimer's disease (AD) is a progressive neurodegener-Alative illness diagnosed clinically by dementia and pathologically by the presence of neuritic plaques in the brain parenchyma and intracellular neurofibrillary tangles. Accumulation of aggregated amyloid- β peptide (A β), the main component of the neuritic plaques,2 is a critical step in the onset of Alzheimer's disease (AD). A β is produced as a 40- or 42-residue peptide following proteolysis of the amyloid- β precursor protein³ and circulates physiologically in the plasma and cerebrospinal fluid.

Convincing evidence of the role of $A\beta$ in AD comes from genetic studies of families that exhibit early onset familial AD (FAD). These studies have identified a number of nucleotide mutations closely linked to $AD^{4,5}$ that either increase $A\beta$ levels or increase the propensity for $A\beta$ aggregation. Furthermore, many of the genetic mutations that cause FAD increase the ratio of $A\beta(1-42)$ to $A\beta(1-40)$. The two additional hydrophobic amino acid residues on $A\beta(1-42)$ substantially increase the propensity for aggregation. It has been shown that neuritic plaques consist overwhelmingly of $A\beta(1-42)^7$ and that $A\beta(1-42)$ forms a greater variety of soluble aggregated species.⁸ Both of these observations, along with other findings, support the critical detrimental role of $A\beta(1-42)$ in AD pathogenesis. In fact, the cerebrospinal fluid (CSF) $A\beta(1-42)$

level is one component of an important biomarker test for AD patients that also includes CSF tau and amyloid imaging.¹⁰

Numerous in vitro studies have shown that unstructured $A\beta$ monomer will undergo noncovalent self-assembly⁶ to form a polydisperse mixture of soluble oligomers^{8,11,12} and/or protofibrils $^{12-14}$ with high β -sheet content 15 and ultimately insoluble fibrils. 16 A β assembly occurs via a nucleationdependent polymerization process. 17 The rate-limiting nucleation step is characterized by a lag phase followed by rapid polymerization and fibril formation. The duration of the lag phase and conformational pathway are variable even in homogeneous aqueous solutions and are sensitive to many conditions, including pH, ionic strength, temperature, and agitation. 13,18-20 A secondary nucleation process may also occur whereby fibril surfaces stimulate new oligomer formation, resulting in the coexistence of both fibrils and oligomers in the same solution.21

Although there remains significant discussion about the impact of soluble and fibrillar forms of $A\beta$, much of the recent emphasis and focus has been on soluble A β aggregates.^{22,23}

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Within this class of soluble aggregated species are oligomers and protofibrils. Numerous studies support the toxicity or biological activity of each species. $^{18,24-27}$ The preparation of either oligomeric or protofibrillar 15,28 A β has been well-documented. These methodologies have been used separately in a variety of studies, but a direct comparison between oligomers and protofibrils has not yet been made.

Multiangle light scattering (MALS) has proven to be a valuable technique for evaluating the self-assembly of multiple amyloid-forming proteins, including $A\beta$, ^{19,29} prions, ³⁰ and α -synuclein. ³¹ Size-exclusion chromatography (SEC) in line with MALS (SEC–MALS) allows separation of polydisperse protein solutions and concurrent size determination. We have previously utilized both oligomer- and protofibril-forming conditions to prepare pools of proinflammatory $A\beta$ for studies of microglial activation. ³² In this study, we utilized light scattering in conjunction with dye binding and microscopy to directly compare the biophysical features of those, as well as other, soluble $A\beta(1-42)$ species.

EXPERIMENTAL PROCEDURES

Preparation of A β **Peptides.** A β (1-42) was obtained from the W. M. Keck Biotechnology Resource Laboratory (Yale School of Medicine, New Haven, CT) in lyophilized form and stored at -20 °C. A β peptide was dissolved in 100% hexafluoroisopropanol (HFIP) (Sigma-Aldrich, St. Louis, MO) to yield a 1 mM A β solution, separated into aliquots in sterile microcentrifuge tubes, and evaporated uncovered at room temperature overnight in a fume hood. The following day, the aliquots were vacuum-centrifuged to remove any residual HFIP and stored in desiccant at -20 °C. $A\beta(1-42)$ protofibrils were prepared as previously described³² by dissolving lyophilized A β (1 mg) in 50 mM NaOH to yield a 2.5 mM A β solution. The solution was then diluted to 250 μ M A β in prefiltered (0.22 μ m) artificial cerebrospinal fluid [aCSF, 15 mM NaHCO₃, 1 mM Na₂HPO₄, 130 mM NaCl, and 3 mM KCl (pH 7.8)]. $A\beta(1-42)$ oligomers were prepared using a slightly modified version of the method of Stine et al.²⁰ in which lyophilized $A\beta$ (1 mg) was dissolved in 100% dimethyl sulfoxide (DMSO) to yield a 5 mM $A\beta$ solution. The solution was then diluted to 100 μ M A β in prefiltered (0.22 μ m) aCSF. The oligomer preparation was incubated for 24 h at 4 °C. Both preparations were centrifuged at 18000g for 10 min with a Beckman-Coulter Microfuge 18 instrument, prior to chromatographic analysis of the supernatants. To discourage protofibril or oligomer formation and increase monomer yields, aliquots of $A\beta(1-42)$ were reconstituted in 6 M guanidine hydrochloride (GuHCl) and 10 mM NH₄OH, centrifuged, and eluted in the selected mobile phase.

Size-Exclusion Chromatography. Supernatants of $A\beta$ (1–42) protofibrils and oligomers were fractionated in line with multiangle light scattering on a Superdex 75 HR 10/30 column (GE Healthcare) using an AKTA FPLC system (GE Healthcare). Prior to injection of $A\beta$, the Superdex 75 column was coated with 2 mg of bovine serum albumin (BSA) taken from a sterile 7.5% fraction V solution (Sigma) to prevent any nonspecific binding of $A\beta$ to the column matrix. Following loading of the sample, $A\beta$ was eluted at a rate of 0.5 mL min⁻¹ in aCSF, and 0.5 mL fractions were collected and immediately placed on ice. In some cases, $A\beta$ was eluted in 50 mM tris(hydroxymethyl)aminomethane hydrochloride (Tris-HCl) (pH 7.8) or aCSF containing 30 mM NaCl (pH 7.8) (aCSF low NaCl). $A\beta$ concentrations were determined by in-line UV

absorbance using an extinction coefficient of 1450 cm $^{-1}$ M $^{-1}$ at 280 nm. A higher-purity (\geq 97%) lyophilized BSA monomer was utilized for molecular weight determination (Sigma). In some cases, A $\beta(1-42)$ monomers were purified by SEC prior to aggregation assays to remove any preexisting aggregates. The elution conditions were the same as those described above, and monomers were recovered in the selected SEC mobile phase. For aggregation assays using SEC-purified monomers, samples were incubated in siliconized tubes at either 37 °C in a water bath using a flotation pad, 20 °C on the benchtop, or 4 °C in the refrigerator.

Multiangle Light Scattering (MALS). MALS data were obtained in line with SEC using a DAWN DSP instrument (Wyatt Technology, Santa Barbara, CA). Samples were analyzed under continuous flow through a K12 quartz cell and illuminated with a helium—neon laser. Light scattering data were obtained every 0.167 mL at fixed-angle detectors 4—16 and plotted using the Debye formalism of the Rayleigh—Debye—Gans approximation^{33,34} for large anisotropic particles (eq 1).

$$R_{\theta}/Kc = M_{w}P(\theta) \tag{1}$$

This equation relates the Rayleigh ratio (scattering intensity) to molecular weight $(M_{\rm w})$ multiplied by a form factor, $P(\theta)$, where c is the solute concentration (grams per milliliter), θ is the scattering angle, K is an optical constant equal to $4\pi^2({\rm d}n/{\rm d}c)^2n_{\rm o}^{\ 2}\lambda_{\rm o}^{\ -4}N_{\rm A}^{\ -1}$, n is the refractive index of the solution, $n_{\rm o}$ is the refractive index of the solution, $n_{\rm o}$ is the refractive index of the solvent, $\lambda_{\rm o}$ is the wavelength of incident light in a vacuum (632.8 nm), and $N_{\rm A}$ is Avogadro's number. $P(\theta)$, which takes into account the destructive interference of scattering by large particles, is the scattering intensity for a large particle divided by the scattering intensity without interference and can be expanded in a power series. The limiting form of the form factor equation as θ approaches zero is shown in eq 2, where $R_{\rm g}^{\ 2}$ is the mean square radius and λ is $\lambda_{\rm o}/n_{\rm o}$.

$$P(\theta) = 1 - (16\pi^2/3\lambda^2)R_g^2 \sin^2(\theta/2)$$
 (2)

Substitution of eq 2 into eq 1 and plotting R_{θ}/Kc versus $\sin^2(\theta/2)$ yield $M_{\rm w}$ as the *y*-intercept and a limiting slope equal to $M_{\rm w}(16\pi^2/3\lambda^2)R_{\rm g}^2$. Debye plots were fit with a linear regression using ${\rm d}n/{\rm d}c$ values for ${\rm A}\beta$ and BSA of 0.186 and 0.190 mL/g, respectively.

Dynamic Light Scattering. Hydrodynamic radius $(R_{\rm H})$ measurements were taken at room temperature with a DynaPro Titan instrument (Wyatt Technology). Samples (35 μ L) were placed directly into a quartz cuvette, and the light scattering intensity was collected at a 90° angle using a 5 s acquisition time. Particle diffusion coefficients were calculated from autocorrelated light intensity data (>25 acquisitions) and converted to $R_{\rm H}$ with the Stokes–Einstein equation. Mean $R_{\rm H}$ values were obtained with Dynamics software (version 6.12.03). Histograms were generated by Dynamics data regularization, and intensity-weighted or mass-weighted mean $R_{\rm H}$ values were derived from the regularized histograms of either percent intensity or mass versus $R_{\rm H}$, respectively.

Thioflavin T Fluorescence. A β solutions were assessed by ThT fluorescence as described previously. A β aliquots were diluted to 5 μ M in aCSF pH 7.8 containing 5 μ M ThT. Fluorescence emission scans (460–520 nm) were conducted on a Cary Eclipse fluorescence spectrophotometer using an excitation wavelength of 450 nm and integrated from 470 to

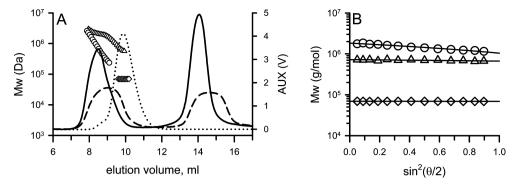


Figure 1. Molecular weight comparison using SEC-MALS. (A) $A\beta(1-42)$ protofibrils and oligomers prepared as described in Experimental Procedures and purified BSA were characterized by SEC-MALS. The absorbance at 280 nm was monitored in line with SEC and converted to voltage (V) using an auxiliary constant calculated for each species. The absorbance-determined concentration traces (right *y*-axis) are shown for $A\beta(1-42)$ protofibrils (—), $A\beta(1-42)$ oligomers (——), and BSA (···), which eluted between 7 and 12 mL. $A\beta(1-42)$ monomers eluted between 13 and 16 mL. M_w values were obtained from Debye plots of MALS data taken from each slice (0.167 mL). The calculated M_w values (left *y*-axis) are plotted for $A\beta(1-42)$ protofibrils (\bigcirc), $A\beta(1-42)$ oligomers (\triangle), and BSA (\diamondsuit). For the M_w plots, every fifth slice is shown for the sake of clarity and distinction between points. (B) Debye plots of the peak slices from panel A for $A\beta(1-42)$ protofibrils (\bigcirc), $A\beta(1-42)$ oligomers (\triangle), and BSA (\diamondsuit) at 8.48, 9.00, and 9.87 mL, respectively.

500 nm to obtain ThT relative fluorescence values. Buffer controls did not show any significant ThT fluorescence in the absence of $A\beta$. All ThT fluorescence values are reported in relative fluorescence units.

Electron Microscopy. SEC-purified $A\beta(1-42)$ protofibrils (10 μ L) were applied to a 200-mesh Formvar-coated copper grid (Ted Pella, Inc.). Samples were allowed to adsorb for 10 min at 20 °C, followed by removal of excess sample solution. Grids were washed three times by placing them sample side down on a droplet of water. Heavy metal staining was conducted by incubation of the grid on a droplet of 2% uranyl acetate (Electron Microscopy Sciences, Hatfield, PA) for 10 min, removal of excess solution, and air drying. Affixed samples were visualized with a JEOL JEM-2000 FX transmission electron microscope operated at 200K eV.

Circular Dichroism. Samples (0.2 mL) were placed into a rectangular quartz cuvette with a 0.1 cm path length, or smaller volumes (15 µL) were measured via a quartz-windowed microsampling disc (Jasco) with a 0.1 cm path length. Spectra were obtained by wavelength scan from 260 to 190 nm using a Jasco J-1500 circular dichroism spectrometer, and four successive wavelength scans were averaged for each A β sample. Buffer control spectra were averaged and subtracted from A β sample spectra, and each resulting point ($[\theta]_{obs}$, degrees) was converted to mean residue ellipticity ($[\theta]$, degrees square centimeter per decimole) with the equation $[\theta]$ = $[\theta]_{\rm obs}(MRW/10lc)$, where MRW is the mean residue molecular weight of A β (1–42) (4514.1 g/mol divided by 42 residues), l is the optical path length (centimeters), and c is the concentration (grams per cubic centimeter). Secondary structure estimates were obtained using the modified Contin method (CON-TINLL) available through the CDPro suite of analysis programs.³⁵ The SMP56 protein reference set was used to estimate percentages of $A\beta(1-42)$ secondary structure from unsmoothed data over the range of 260-196 nm.

RESULTS

Multiangle Light Scattering Analysis of $A\beta$ Protofibrils and Oligomers. Equivalent aliquots (1 mg) of HFIP-treated and lyophilized dry $A\beta(1-42)$ were brought into solution separately using previously reported protofibril-forming 14,28,32 and oligomer-forming 20 procedures and conditions (described

in Experimental Procedures). Rather than phenol red-free F-12 cell culture medium, the $A\beta$ species were allowed to form in aCSF. We have recently described A β protofibril formation in aCSF,³⁶ a carbonate/phosphate buffering system (pH 7.8) similar in composition to phenol red-free F-12, but without glucose, Ca²⁺, and Mg²⁺. For protofibrils, the solution (1 mL) was centrifuged immediately, without additional incubation time, at 18000g for 10 min and the supernatant characterized by SEC-MALS. Elution of the $A\beta(1-42)$ protofibril solution on Superdex 75 yielded a distinct two-population separation of protofibrils and monomers as monitored by 280 nm absorbance (Figure 1A, solid line). Light scattering data were collected continuously during the elution at 13 fixed angles and analyzed in 0.0167 mL data fractions or "slices". Each slice was analyzed with a Debye plot shown in Figure 1B. A linear fit of the angledependent light scattering intensity for each elution slice allowed the determination of $M_{\rm w}$ from the y-intercept and rootmean-square radius (R_{σ}) from the slope. $M_{\rm w}$ values taken from the full width at half-maximum of the protofibril concentration peak (7.87–9.13 mL) ranged from 2600 kDa (576 monomers) at the leading side to 228 kDa (50 monomers) for the trailing side (Figure 1, circles). Although the protofibril fraction eluted in the void volume, a significant degree of size separation was observed. This is the first SEC-MALS analysis of this frequently reported method of protofibril preparation, and a large degree of size polydispersity was evident rather than a homogeneous population of similarly sized aggregates. The pattern of protofibril elution was a smooth decreasing gradient of sizes with an even distribution of M_w values across the protofibril peak. For validation purposes, SEC-MALS was used to characterize a purified BSA/aCSF solution. BSA (Figure 1, dotted line) eluted in the included volume of the Superdex 75 column (fractionation range of 3-70 kDa), and the measured mean BSA M_w ± standard deviation (SD) from 44 slices (Figure 1, diamonds) across the concentration peak was 69.6 \pm 0.6 kDa.

We also utilized a commonly used and well-described method for preparing oligomeric $A\beta$, which entails reconstitution of the dry $A\beta(1-42)$ peptide in DMSO followed by dilution into cold F12 medium and incubation for 24 h at 4 °C. For this investigation, phenol red-free F-12 medium was replaced with aCSF. The same SEC–MALS analysis was

conducted on an A β (1-42) oligomer preparation. Because oligomers are formed at 100 μ M rather than 250 μ M for the protofibrils, twice the volume was loaded onto the column so that total amounts of $A\beta$ were closer to equivalent. The oligomers began to elute at the same time as the protofibrils, but the larger injection volume resulted in some peak broadening. The oligomer elution also displayed two major populations (monomers and higher-molecular weight species) with no evidence of intermediate species between the two peaks. The $M_{\rm w}$ range for $A\beta(1-42)$ oligomers taken from the full width at half-maximum of the void volume concentration peak (7.95-9.93 mL) ranged from 2967 kDa (657 monomers) at the leading edge of the concentration peak to 588 kDa (130 monomers) at the trailing side edge of the concentration peak (Figure 1, triangles). Debye plots of slices taken from the protofibril, oligomer, and BSA concentration peaks in Figure 1A are shown in Figure 1B and represent the largest population of each protein eluted from the column. M_w values determined from the three plots were 69 kDa for BSA, 723 kDa for A β (1– 42) protofibrils, and 1718 kDa for $A\beta(1-42)$ oligomers.

 $R_{\rm g}$ is a size parameter based on the center of mass and internal distribution of individual mass elements for a particular particle. z-average $R_{\rm g}$ values were obtained from the same protofibril and oligomer SEC–MALS elutions shown in Figure 1, and these values corresponded directly with the $M_{\rm w}$ measurements in each slice. In Figure 2, the 90° detector

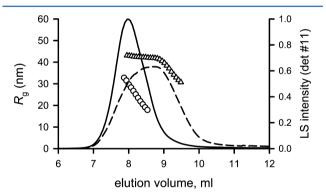


Figure 2. In-line $R_{\rm g}$ determination of ${\rm A}\beta(1{\rm -}42)$ preparations. $R_{\rm g}$ values were obtained from Debye plots constructed during the SEC–MALS analysis in Figure 1. The light scattering intensity (right *y*-axis) from the 90° detector (detector 11) during the elution is shown for ${\rm A}\beta(1{\rm -}42)$ protofibrils (—) and oligomers (——). $R_{\rm g}$ values plotted above (left *y*-axis) represent 50% of the maximal concentration trace for the leading edge to 50% of the maximal light scattering intensity trace for the trailing edge.

light scattering trace is shown rather than the concentration trace. Values of $R_{\rm g}$ for ${\rm A}\beta(1{\rm -}42)$ protofibrils were much smaller than those of the ${\rm A}\beta(1{\rm -}42)$ oligomers. Protofibrils ranged from 33 nm (leading edge) to 18 nm (trailing edge) (z-averaged mean of 25 nm), while oligomers displayed a more homogeneous population from 40 to 43 nm before decreasing in size from 40 to 30 nm across the trailing side of the light scattering peak (z-averaged mean of 40 nm) (Figure 2).

A second SEC–MALS analysis was conducted in which the protofibril and oligomer preparations were analyzed on the same day in successive SEC–MALS runs. This set of experiments revealed that protofibril preparations were very reproducible. Identical aggregation conditions produced similar protofibril $M_{\rm w}$ distributions (Figure 3) ranging from 2418000 Da (leading edge) to 296600 Da (trailing edge). Oligomer

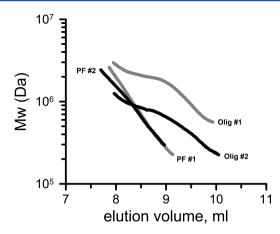


Figure 3. Protofibril preparation yields a consistent $M_{\rm w}$ range, while the oligomer preparation is more variable. A $\beta(1-42)$ protofibrils and oligomers (each 1 mg dry weight) were prepared as described in Experimental Procedures and the legend of Figure 1. The oligomer preparation was initiated the previous day to allow successive SEC–MALS analyses to be conducted. The in-line SEC–MALS-determined $M_{\rm w}$ range during elution is shown for the first preparation of protofibrils (PF #1, gray line) and oligomers (Olig #1, gray line) described in Figure 1. The $M_{\rm w}$ ranges obtained during the repeat experiment are shown as black lines for protofibrils (PF #2) and oligomers (Olig #2).

sizes, however, were much more variable between preparations, and overall $M_{\rm w}$ values for the second preparation differed quite a bit from those of the first preparation (Figure 3). The oligomer M_w range in the second preparation was 1255 kDa (278 monomers) at the leading edge and 226 kDa (50 monomers) at the trailing edge. The variability in the M_w values between different oligomer preparations may be explained by the variability in nucleation lag time during the 24 h incubation time utilized for their formation. Protofibrils, on the other hand, are formed rapidly at the higher concentration under conditions outlined in Experimental Procedures, leaving less variability in lag time. Despite the differences in oligomer M_w between preparations, R_{σ} values for the second protofibril and oligomer preparations were similar to those of the initial preparations. $A\beta(1-42)$ protofibrils ranged from 31 to 17 nm across the light scattering peak (z-averaged mean of 25 nm) (Figure 4), while $A\beta(1-42)$ oligomers ranged from 41 to 22 nm (z-averaged mean of 35 nm).

Dynamic Light Scattering Analysis. During the second round of analysis, dynamic light scattering (DLS) measurements of the collected fractions were included to obtain further information about the protofibril and oligomer preparations. Although the oligomers displayed rms radius values somewhat larger than those of protofibrils, the mean regularized hydrodynamic radius (R_H) obtained from intensity-weighted histograms for the oligomers (Figure 4, triangles) was similar to, but slightly lower than, that of the protofibrils (Figure 4, circles). R_H values for three fractions across the protofibril peak were 34 nm (8.0 mL), 30 nm (8.5 mL), and 22 nm (9.0 mL), while fractions across the oligomer peak were 31 nm (8.5 mL), 26 nm (9.0 mL), and 19 nm (9.5 mL). Averaging of the fraction R_H through the peak for protofibrils and oligomers yielded values of 29 and 25 nm, respectively. The unregularized mean R_H typically contains contributions from multiple particle populations, including trace amounts of very large particles and the buffer (solvent) component. The use of regularized histograms allows separation of particle populations and a

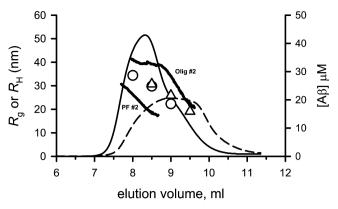


Figure 4. Comparison of $R_{\rm g}$ and $R_{\rm H}$ for ${\rm A}\beta(1{\rm -}42)$ protofibrils and oligomers. $R_{\rm g}$ values were obtained in line during the second SEC–MALS analysis described in Figure 3. A UV absorbance trace was used to determine the ${\rm A}\beta$ concentration in line (right y-axis) for protofibrils (—) and oligomers (——). $R_{\rm g}$ values (left y-axis) are shown from 50% of the maximal concentration trace for the leading edge to 50% of the maximal light scattering intensity trace for the trailing edge for protofibrils (PF #2) and oligomers (Olig #2). $R_{\rm H}$ values (left y-axis) were determined in the fractions collected (#17, 8.0 mL; #18, 8.5 mL; #19, 9.0 mL; #20, 9.5 mL) during elution of the protofibrils (O) and oligomers (Δ).

better determination of $A\beta$ protofibril or oligomer size. Particles that were detected by the instrument with an $R_{\rm H}$ of >1000 nm were not considered. $R_{\rm H}$ (DLS) was larger than $R_{\rm g}$ (MALS) for protofibrils but smaller for the oligomers in this particular experiment, but that was not the norm. The ratio between the two parameters can signify a particular conformation for a particle and is discussed in more detail below. In all, multiple preparations of $A\beta(1-42)$ protofibrils (n=4) and oligomers (n=3) were analyzed, and the summary of the light scattering data is shown in Table 1. Although both $A\beta$

Table 1. Mean Sizes for $A\beta(1-42)$ Protofibrils and Oligomers^a

sample	$M_{ m w} \pm { m SD}$	$R_{\rm g} \pm { m SD}$	$R_{\rm H} \pm { m SD}$
protofibrils	$884 \pm 302 (4)$	$25.5 \pm 1.8 (4)$	$23.6 \pm 6.3 (3)$
oligomers	1317 + 595(3)	37.1 + 2.5(3)	31.3 + 5.5(3)

"Values obtained from the full width at half-maximum of the concentration peak (n experiments). Units of kilodaltons for $M_{\rm w}$ and nanometers for $R_{\rm g}$ and $R_{\rm H}$. $R_{\rm H}$ was determined from the regularized histogram of each fraction in the upper third of the SEC concentration peak and averaged together to yield an overall $R_{\rm H}$ for each protofibril or oligomer preparation.

preparations are polydisperse with respect to size, mean measurements were obtained over the multiple analyses and presented with the standard deviation between the analyses in Table 1. Overall, $M_{\rm w}$, $R_{\rm g}$, and $R_{\rm H}$ values were smaller for protofibrils than for oligomers. The combination of light scattering techniques provided a clearer picture of protofibril and oligomer sizes and indicated that both species were soluble, but relatively large, and contained somewhere between 50 and 660 monomer units. Protofibrils were routinely smaller in size and had a molecular weight lower than that of oligomers when adhering to the preparation conditions described in this report.

Structure and Morphology Analysis. Thioflavin T (ThT) binding and fluorescence is a commonly used method for monitoring amyloid formation.³⁷ Dye binding and subsequent fluorescence emission at 482 nm correlate well

with $A\beta$ aggregation and formation of β -sheet secondary structure. While ThT has been used frequently to assess protofibrils, this is not typical with oligomer preparations. We have previously demonstrated that SEC-isolated $A\beta(1-42)$ protofibrils exhibit significant ThT fluorescence, albeit much less than an equimolar solution of $A\beta(1-42)$ fibrils, while SEC-isolated $A\beta(1-42)$ monomers do not generate ThT fluorescence. ThT fluorescence. ThT fluorescence after SEC-MALS analysis were directly compared for ThT fluorescence. Fractions collected from a third SEC-MALS analysis of protofibrils and oligomers were assessed, and similar levels of ThT fluorescence were observed for all fractions through both protofibril and oligomer elution peaks (Figure 5). The final $A\beta$ concentration was maintained at

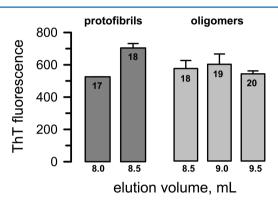


Figure 5. $A\beta(1-42)$ protofibrils and oligomers have equivalent levels of thioflavin T binding and fluorescence. Aliquots of $A\beta(1-42)$ protofibril and oligomer fractions collected after SEC-MALS analysis were diluted to 5 μ M in aCSF containing 5 μ M ThT, and fluorescence emission was measured and plotted as described in Experimental Procedures. Error bars represent the standard error of n=3 ThT measurements. The numbers shown within the data bars indicate the collected fractions during elution and correspond to the fractions and elution volumes in the legend of Figure 4.

5 μ M in all fluorescence measurements. The ThT fluorescence findings indicated that oligomers contain elements of β -sheet and amyloid structure similar to that of protofibrils.

Morphological differences between isolated $A\beta(1-42)$ protofibrils and oligomers were examined using transmission electron microscopy (TEM). We, and others, have previously used TEM to characterize $A\beta(1-42)$ protofibrils. 14,32,38 Protofribrils are typically curvilinear structures <200 nm in length. High-magnification (59000x) TEM images of the protofibril fraction collected at 8.5 mL in Figure 3 revealed the classical morphology. Numerous small flexible $A\beta(1-42)$ structures were observed (Figure 6A), and length analysis produced a mean value of 78 ± 35 nm (standard deviation). The analysis included only unambiguous protofibril structures and excluded the smaller species (<20 nm). The particular protofibril and oligomer fractions analyzed and depicted in the TEM image in Figure 6A were obtained following the SEC-MALS analysis presented in Figure 3 (PF #2, Olig #2). The corresponding M_w range for the protofibril fraction was 654-296 kDa with an average $R_{\rm H}$ of 30 nm. The oligomer fraction displayed none of the morphological characteristics of protofibrils despite possessing similar size features ($M_{\rm w}$, 651– 410 kDa; $R_{\rm H}$, 26 nm). The A β (1–42) oligomers were globular spheroid shapes with a fair amount of polydispersity in crosssectional distance (Figure 6B).

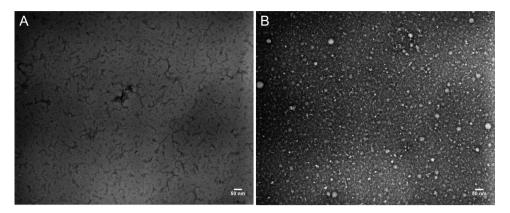


Figure 6. Morphological comparison between $A\beta(1-42)$ protofibrils and oligomers. TEM images were obtained from the second $A\beta(1-42)$ protofibril (A) and oligomer (B) preparation described in the legends of Figures 3 and 4. Samples were prepared as described from fractions collected after SEC-MALS analysis, and images were captured at 59000× magnification (scale bar of 50 nm). Length analysis was conducted manually utilizing the scale bar length to prepare 50, 100, and 150 nm increment bars to determine protofibril lengths in four quadrants of the image (total n = 106).

Analysis of Soluble $A\beta(1-42)$ Species Formed from SEC-Purified Monomers. Every effort was made in the studies described above to precisely replicate the conventional protofibril and oligomer preparations from dry peptide films that have been reported in the literature. The results in this study indicate that high A β concentrations at room temperature favor rapid protofibril formation. However, working with stored dry $A\beta(1-42)$ peptide samples without further purification may permit formation of preexisting structures that influence the ensuing aggregation process. For these studies, $A\beta(1-42)$ was reconstituted in 6 M GuHCl containing 10 mM NH₄OH to suppress rapid protofibril formation and increase monomer yields. Monomer fractions were obtained from the Superdex 75 included volume, immediately placed on ice, and promptly used for aggregation studies. The A β (1–42) concentration of 40 μ M was chosen so that 0.5 mL (90 μ g A β) could be loaded for SEC-MALS and postelution evaluation. This amount was the realistic minimum for sample detection and analysis. SECpurified monomeric $A\beta(1-42)$ exhibited no ThT fluorescence (Figure 7A), and subsequent SEC repurification yielded no material in the Superdex 75 void volume (Figure 7B). SECpurified $A\beta(1-42)$ monomer incubated at 37 °C in aCSF demonstrated a lag phase of >4 h but then aggregated rapidly overnight (21 h). All of the ThT-fluorescent material was insoluble and was removed from the solution after centrifugation at 18000g for 10 min (Figure 7A). SEC analysis of the supernatant showed negligible material in the void volume and little remaining monomer in the included volume (Figure 7B), indicating a rapid progression from monomer to insoluble fibrils. Previous studies have shown a significant dependence on ionic strength for insoluble fibril formation.¹⁹ Therefore, aCSF pH 7.8 was prepared with 30 mM NaCl, rather than 130 mM (aCSF low NaCl). $A\beta(1-42)$ monomers purified by SEC in this buffer (40 μ M) were incubated at 37 °C overnight and demonstrated a longer lag phase [>8 h (data not shown)] yet still progressed to a significant ThT fluorescence overnight (25 h); 90% of the ThT-positive A β (1–42) aggregates remained soluble after centrifugation (Figure 7A), and a substantial peak was observed in the SEC void volume (Figure 7B). Tris-HCl has frequently been used as a buffer for $A\beta$ studies, 14,15,19 and it was employed in this study as a comparison to aCSF. A β (1–42) monomers purified via SEC in 50 mM Tris-HCl pH 7.8 (40 μ M) were incubated at 37 °C

overnight. The progression of aggregation for $A\beta(1-42)$ in Tris-HCl was slightly slower than in aCSF low NaCl, yet a significantly lower percentage of the ThT-positive material (9%) remained soluble in the solution after centrifugation. SEC analysis matched this observation with less $A\beta(1-42)$ overall eluted from the column.

Despite the loss of $A\beta$ after centrifugation, a notable void volume peak was still present for evaluation. MALS analysis in line with SEC provided $M_{\rm w}$ and $R_{\rm g}$ values for the soluble $A\beta(1-$ 42) aggregates formed after a 37 $^{\circ}$ C incubation. $M_{\rm w}$ values for aggregates formed in aCSF low NaCl ranged from 3394 to 2961 kDa (mean of 3044 kDa, 674 monomers), while aggregates formed in Tris-HCl were notably smaller and ranged from 2133 to 1386 kDa (mean of 1533 kDa, 340 monomers) (Figure 7C). This observation was somewhat surprising in that much of the $A\beta(1-42)/Tris-HCl$ solution was insoluble, which was indicative of fibril formation, yet the supernatant contained lower- $M_{\rm w}$ species. $R_{\rm g}$ values corresponded to the $M_{\rm w}$ determinations and ranged from 59 to 58 nm (mean of 59 nm) for aggregates formed in aCSF low NaCl and 43-37 nm (mean 39 nm) for aggregates formed in Tris-HCl (Figure 7D). DLS-generated $R_{\rm H}$ values for the postelution peak fractions were slightly harder to interpret because of polydispersity in the aCSF low NaCl aggregate solution. The percent intensity $R_{\rm H}$, which can be skewed by larger species, was 139 nm, yet the largest percent mass population (83%) was found at 60 nm, similar to the $R_{\rm g}$ value (data not shown). $R_{\rm H}$ values for aggregates formed in Tris-HCl indicated less polydispersity with the greatest population based on the percent intensity at 25 nm and based on the percent mass at 17 nm (data not shown). The differences in morphology between the SEC-isolated A β (1–42) soluble aggregates formed in aCSF low NaCl pH 7.8 and 50 mM Tris-HCl pH 7.8 were quite striking. TEM images revealed long (>1 μ m), slender protofilament structures in aCSF low NaCl (Figure 7E), while those in Tris-HCl were curvilinear protofibril structures 50-100 nm in length (Figure 7F). The observation of protofibrils in a Tris-HCl solution containing primarily insoluble fibrils demonstrated the coexistence of these two species prior to centrifugation and SEC isolation and emphasized the polydispersity that can occur in A β aggregation. The isolated $A\beta(1-42)$ protofilaments from the aCSF low NaCl solution were fairly homogeneous with the peak SEC

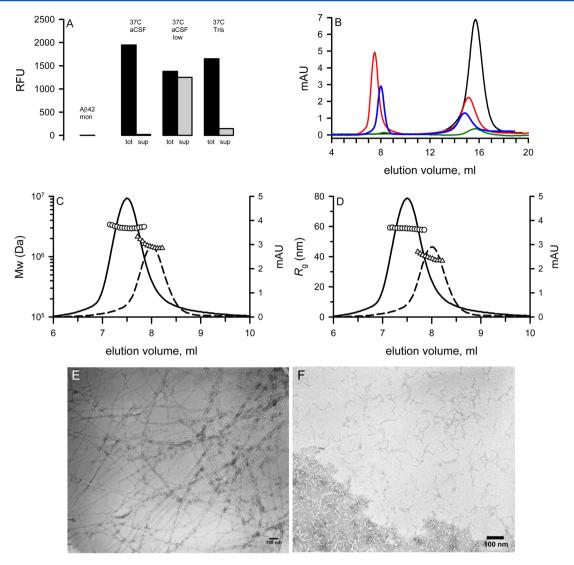


Figure 7. Ionic strength and buffer type influence soluble $A\beta(1-42)$ aggregate formation by SEC-purified $A\beta(1-42)$ monomers. (A) Following SEC purification of $A\beta(1-42)$ monomers in differing buffers, ThT fluorescence measurements were taken at periodic times before ($A\beta42$ mon), during, and after incubation at 37 °C overnight. All solutions had an $A\beta$ concentration of 40 μM. The final fluorescence time points are depicted in the plot and represent 21 h for monomers purified in aCSF pH 7.8, 25 h for monomers purified in aCSF low NaCl pH 7.8 (aCSF low), and 32 h for monomers purified in 50 mM Tris-HCl pH 7.8 (Tris). ThT fluorescence measurements were obtained from the solutions before (total), and in the supernatant (sup) after, centrifugation at 18000g for 10 min. At each time point and experimental step, ThT fluorescence was assessed in the absence of $A\beta$. These values ranged from 28 to 116 RFU and were subtracted from the $A\beta$ /ThT values. (B) Supernatants (0.5 mL) from the solutions in panel A were repurified via SEC in their respective buffers, and the chromatograms obtained by 280 nm absorbance (mAU) are shown for $A\beta(1-42)$ monomers in aCSF (black line) and the aCSF (green line), aCSF low NaCl (red line), and Tris (blue line) reactions. (C and D) SEC-MALS analysis was conducted on the soluble aggregates eluted in the Superdex 75 void volume. M_w (C) and R_g (D) values were determined every 0.167 mL for soluble $A\beta(1-42)$ aggregates formed in aCSF low NaCl (O) and Tris (Δ). Every fourth value is shown for the sake of clarity and distinction between points. The 280 nm absorbance traces from panel B are reproduced for aCSF low NaCl (—) and Tris (—). (E and F) TEM images were obtained from the peak void volume fractions collected from the elutions shown in panels C and D. Images were captured at 35000× (E, aCSF low NaCl) and 59000× (F, Tris) magnifications (scale bars of 100 nm).

void volume fraction. Their preparation was quite reproducible as a second experiment produced similar long slender protofilament structures with a similar mean $M_{\rm w}$ and $R_{\rm g}$ (2896 kDa and 57 nm, respectively) (data not shown).

It was clear in our studies utilizing SEC-purified $A\beta(1-42)$ monomers in aCSF low NaCl pH 7.8 that elongated, yet still quite soluble, protofilaments formed rapidly following nucleation. The rapid kinetics hindered observation of protofibrillar or oligomeric species. Because $A\beta$ aggregation is temperature-dependent, additional incubations were conducted at 20 and 4 °C, and the aggregation kinetics were compared to

those of the 37 °C incubation described above. Using ThT fluorescence as an aggregation marker, it was clear that nucleation of SEC-purified A β (1–42) monomers (40 μ M) was keenly dependent on temperature, suggesting a significant activation energy barrier for aggregation initiation (Figure 8A). Following nucleation of the samples at the different temperatures, the solution supernatants after centrifugation were analyzed by ThT fluorescence and SEC. The 20 and 4 °C samples retained significant soluble ThT-positive material (92 and 68%, respectively) after centrifugation that was observed in the SEC void volume (Figure 8B). $M_{\rm w}$ and $R_{\rm g}$ values ranged

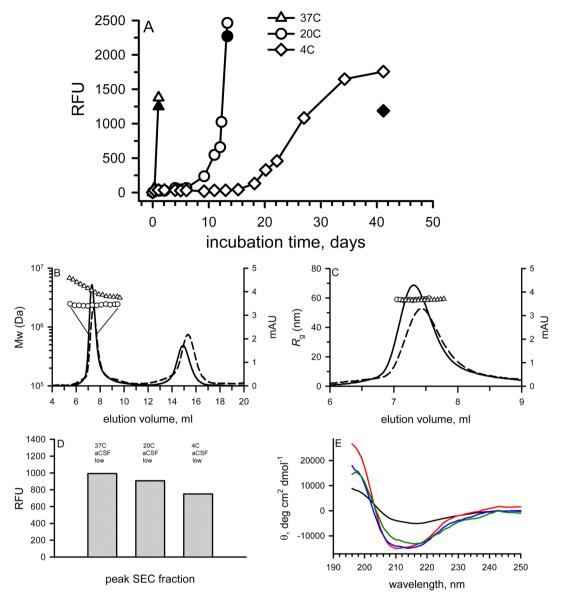


Figure 8. $A\beta(1-42)$ aggregation is highly temperature-dependent in aCSF low NaCl, yet similar soluble species are formed. (A) Solutions of SEC-purified $A\beta(1-42)$ monomers (40 μ M) in aCSF low NaCl were incubated at 4 (\triangle), 20 (\bigcirc), and 37 °C (\diamondsuit). ThT fluorescence measurements were taken at periodic times. At the final aggregation time points for each temperature, the solutions were centrifuged at 18000g for 10 min, and the supernatants were probed for ThT fluorescence (filled symbols). (B and C) Centrifugation supernatants (0.5 mL) from the solutions in panel A were repurified on SEC in aCSF low NaCl. The 280 nm absorbance elution traces for the 37 °C elution are shown in Figure 7B (---), while the 20 (--) and 4 °C (---) aggregation solutions are shown above (right *y*-axis). M_w (B) and R_g (C) values (left *y*-axes) were determined in line with SEC–MALS for each 0.167 mL fraction (slice) through the void volume peaks for 20 (\bigcirc) and 4 °C (\triangle). Because the complete elution trace is shown in panel B, the M_w values, determined from the full width at half-maximum of each peak, have been expanded. (D) ThT fluorescence measurements of the peak void volume fractions eluted after SEC at 1.04 (37 °C incubation), 13.3 (20 °C incubation), and 41.2 (4 °C incubation) days were obtained. Because of the low $A\beta(1-42)$ concentrations, samples (35 μ L) were mixed in a 1:1 ratio with 10 μ M ThT to enhance the fluorescence signal. (E) Peak void volume SEC fractions and the precentrifuge solution from the 37 °C aCSF reaction depicted in Figure 7A (second set of bars) (black line) were analyzed by CD. Spectra were obtained as described in Experimental Procedures for SEC-isolated peak void volume fractions from the 37 °C (red line), 20 °C (blue line), and 4 °C (green line) reactions in aCSF low NaCl. MRW was calculated on the basis of 40 μ M A β (1-42) for 37 °C in aCSF and 7, 6, and 5 μ M for 37, 20, and 4 °C aggregation reactions in aCSF low NaCl, respectively.

from 2478 to 2281 kDa (mean of 2357 kDa, 522 monomers) and from 59 to 58 nm (mean of 59 nm), respectively, for the 20 $^{\circ}$ C reaction and from 6779 to 3140 kDa (mean of 4324 kDa, 958 monomers) and from 59 to 58 nm (mean of 58 nm), respectively, for the 4 $^{\circ}$ C reaction (Figure 8B,C). Peak fractions collected from the void volume of the three SEC elutions all exhibited similar ThT fluorescence levels (Figure 8D), the only differences corresponding to the slightly different $A\beta$

concentrations (37 °C, fraction 16, 7 μ M; 20 °C, fraction 16, 6 μ M; 4 °C, fraction 16, 5 μ M).

Circular dichroism (CD) was utilized to further probe the secondary structure of the SEC-isolated soluble $A\beta(1-42)$ protofilaments. The precentrifuge insoluble fibrils that were obtained after overnight incubation of SEC-purified $A\beta(1-42)$ monomers in aCSF (described in Figure 7A) had significant amounts of β -sheet structural elements. Secondary structure estimates using CONTINLL in the CDPro software package

determined 8% helix, 63% sheet/turn, and 29% coil for the structures formed in the higher-NaCl aCSF buffer (Figure 8E). Surprisingly, the soluble A β (1–42) protofilaments formed in aCSF low NaCl and isolated by SEC contained significant α -helical structure along with the β -sheet (Figure 8E). The soluble species formed most rapidly at 37 °C exhibited the most helical structure (45% helix, 25% sheet/turn, and 30% coil) with the slower-forming soluble species at 20 °C (34% helix, 34% sheet/turn, and 32% coil) and 4 °C (24% helix, 54% sheet/turn, and 22% coil) containing less. The molar ellipticities were different because of the higher solution concentration of the fibrils formed in aCSF (40 μ M) compared to that of the SEC-isolated protofilaments formed in aCSF low NaCl (4–7 μ M).

DISCUSSION

Two of the most frequently used in vitro $A\beta$ aggregate preparations for biological studies have been reported to result in the formation of either "protofibrils" or "oligomers". A β protofibrils were first observed more than 15 years ago and were characterized as precursors along the fibril-forming pathway. 14,39 The conditions have been optimized over time to increase protofibril yields and minimize polydispersity. 40,41 An important factor involved solvation of the A β under alkaline conditions using either NH₄OH or NaOH. Typically, A\beta is synthesized as a trifluoroacetic acid salt, which when solubilized results in an acidic solution. Transfer into buffers to conduct studies at neutral pH required crossing the A β isoelectric point (pI 5.5). The alkaline method avoided this step and significantly slowed the introduction of polydispersity and loss of solubility experienced at the pI. Because of the widespread interest in soluble A β species, detailed procedures for the preparation of protofibrillar $A\beta$ have continued to be refined and reported.²⁸ Because many studies are aimed at investigating the biological effects of $A\beta$ protofibrils, we have recently reported the preparation and characterization of $A\beta(1-42)$ protofibrils in antibiotic-supplemented F-12 cell culture medium³² and a carbonate- and phosphate-based aCSF

SEC isolation of the protofibrils has allowed further structural and morphological analysis. As determined by circular dichroism (CD) and ThT binding and fluorescence, $A\beta$ protofibrils possess significant β -structure, albeit to a lesser extent than $A\beta$ fibrils. Microscopy analysis has shown curvilinear structures <200 nm in length. However, true $M_{\rm w}$ determination for this specific type of protofibril preparation has not been reported. In this study, we used SEC–MALS to provide the first characterization of $M_{\rm w}$ and $R_{\rm g}$ for rapidly formed $A\beta(1-42)$ protofibrils and compare the findings with those obtained for $A\beta(1-42)$ oligomers.

 $A\beta(1-42)$ protofibrils eluted in the void volume of a Superdex 75 column (fractionation range of 3–70 kDa), yet size separation was observed as the protofibril $M_{\rm w}$ values averaged 2119 kDa at the leading edge of the concentration peak and 307 kDa at the trailing edge (2600 kDa maximum, 208 kDa minimum) over four separate SEC–MALS analyses. The average $M_{\rm w}$ range represents an equivalent protofibril size in monomer units from 68 to 469 monomers. Protofibrils are exceedingly soluble and relatively small compared to fibrils, but they do not exist as a discrete homogeneous population. In fact, there was significant polydispersity in the protofibril size distribution. Interestingly, no species were observed between protofibrils and $A\beta(1-42)$ monomers (4514 Da). $A\beta(1-42)$

protofibrils are formed rapidly (0.5–2 h) after addition of aCSF to the NaOH-solubilized A β peptide. A much longer incubation time is necessary for the formation of A β (1–40) protofibrils. Rosenberry and colleagues previously used SEC–MALS to examine soluble A β (1–40) protofibrils prepared from the SEC-purified monomer under vigorous agitation conditions. Protofibril sizes in the earlier study were larger ($M_{\rm w}=7000-32000~{\rm kDa}$; $R_{\rm g}=50-120~{\rm nm}$) than those observed in the study presented here and likely represented protofibrils at later stages in the aggregation process because of the mechanical agitation.

Our SEC-MALS analysis of the $A\beta(1-42)$ oligomer preparation was somewhat surprising in that, in multiple preparations, oligomers were larger than protofibrils. These results are consistent with those of a previous light scattering analysis of a soluble oligomeric preparation termed an amyloid- β -derived diffusible ligands (ADDLs) oligomer preparation.²⁹ Hepler et al. used SEC-MALS to characterize the size of ADDLs in solution and observed a polydisperse mixture of oligomers ranging in size from nearly 1000 kDa at the leading edge of the peak to 150 kDa at the trailing edge. The method used for the preparation of ADDLs in the previous study is the basis for the oligomer preparation we characterize in this study and has been rigorously detailed in a separate report. 20 The $M_{\rm w}$ values obtained for $A\beta(1-42)$ oligomers in our current study averaged 2379 kDa at the leading edge and 611 kDa at the trailing edge, larger than those in the earlier investigation. Two factors likely played a role in this size difference. One, we observed that oligomer formation and polymerization were more variable than protofibril formation and required multiple analyses to obtain a median size. While the oligomer solution always contained higher- M_w species, the M_w values could fluctuate 2–3-fold. Two, different A β dn/dc values were used in the two studies. Hepler et al. used an empirically derived value of 0.241 mL g⁻¹, while in the study presented here, the general protein value of 0.186 mL g⁻¹ was incorporated. The latter value was also used in previous A β studies. ¹⁹ Incorporation of the higher dn/dc value would bring our oligomer M_w values closer in line with those of Hepler et al.

While $A\beta(1-42)$ oligomers were typically larger than protofibrils in terms of $M_{\rm w}$, $R_{\rm e}$, and $R_{\rm H}$, the two species showed similarities in ThT binding and fluorescence and extended conformation. However, there was a notable difference in morphology. The TEM image in Figure 6 showing the spherical globular morphology of an SEC-isolated oligomer preparation is similar to that observed previously by AFM. 20,29 Morphologically, it is not clear why Aeta oligomers are consistently characterized as globular spheroid by microscopy yet in solution are observed as higher-molecular weight aggregates with an extended conformation. Oligomeric A β has typically been characterized by denaturing (SDS) or nondenaturing (native) gel electrophoresis with a focus on dimeric, trimeric, tetrameric, and other low-molecular weight species. 18,20,42 However, significant amounts of these small oligomers were not detected by SEC-MALS between the higher-molecular weight species and the monomers, including the leading edge of the monomer peak. There is a possibility that these small oligomeric $A\beta$ species exist at very low levels and are difficult to detect in solution. One conclusion of this study is that, in most conventional preparations starting with dry $A\beta(1-42)$ peptide, both protofibrils and oligomers may coexist and possess size and structural similarities. However, one area of divergence may be solvent-exposed aggregate surfaces. This difference would likely alter the surface properties

of each species, greatly impact adsorption to various substrates, and lead to divergent apparent morphologies using microscopy. This is consistent with the many observations that $A\beta$ aggregates typically have some degree of structural polydispersity and are rarely homogeneous.

One of the concerns in the studies of aggregation-prone peptides and/or proteins is the starting material. Routinely used preparations involve reconstitution from a dry peptide at high concentrations. Even though precautions are typically taken such as HFIP pretreatment and high-alkaline reconstitution, some preexisting aggregate formation may be possible during synthesis, dissolution, lyophilization, drying, and storage of the $A\beta(1-42)$ stocks. These aggregates may persist even after HFIP or DMSO treatment. Our studies indicate that $A\beta(1-42)$ protofibril formation is rapid and favorable at high concentrations. However, at lower concentrations and after further purification of $A\beta(1-42)$ monomers by SEC, the lag phase prior to nucleation was extended. Once nucleation had occurred, the aggregation progressed quickly to long slender soluble protofilaments at lower ionic strengths. These species rapidly achieved $M_{\rm w}$ and $R_{\rm g}$ values of 3000 kDa (665 monomers) and 60 nm, respectively. A higher ionic strength favored high-mass, insoluble fibril formation, which hindered some solution analyses. Interestingly, no globular or spherical oligomeric species were observed in the aggregation reaction mixtures that started with SEC-purified $A\beta(1-42)$ monomers. These species may be present prior to the onset of ThT binding and fluorescence.

The experimental R_g/R_H ratio can provide conformational information about polymeric species. R_g/R_H values of 0.78 for spheres (globular proteins) and 1.30° for flexible Gaussian chains are typical. ⁴³ Both $A\beta(1-42)$ protofibrils and oligomers had R_{o}/R_{H} ratios of >0.78 indicative of an extended conformation. The average $R_{\rm g}/R_{\rm H}$ value obtained over multiple separate experiments was 1.08 ± 0.30 (standard deviation) for protofibrils and 1.19 ± 0.22 for oligomers. The standard deviation provided evidence of some conformational polydispersity in both preparations and may represent the presence of alternate structures. A more detailed analysis of fractional R_{σ} / R_H through each protofibril and oligomer peak yielded values of 1.09 ± 0.31 (standard deviation) and 1.11 ± 0.16 for protofibrils and oligomers, respectively. $R_{\rm g}/R_{\rm H}$ comparisons were more difficult with protofilaments formed from SECpurified A β (1–42) monomers because of polydispersity in the $R_{\rm H}$ measurements.

 $A\beta$ protofibrils and oligomers have been shown to elicit biological responses, including a proinflammatory response in immune cells. ^{32,44–48} The absolute molecular details of soluble $A\beta$ aggregates such as protofibrils and oligomers that render them neurotoxic or proinflammatory remain to be determined. These properties will likely include size, secondary structure, overall conformation, and specific exposed surfaces. *In vitro* studies, so valuable for elucidating these fine details, will ultimately need to be validated *in vivo*.

CONCLUSION

This study is the first direct comparison of the biophysical properties of $A\beta(1-42)$ protofibrils and oligomers using their respective conventional preparations. These two species are part of a broader class of soluble aggregated $A\beta$ assemblies believed to play a critical role in AD pathogenesis. Although they are very soluble, we found that both species exist as highermolecular weight aggregates in solution, with the oligomers

actually being larger than the protofibrils. There were notable biophysical similarities and key morphological differences between the two, both of which differed from soluble protofilaments formed from SEC-isolated monomers. This study provides important information about the biophysical properties of $A\beta$, and this knowledge will support future studies aimed at understanding how the properties contribute to the neurodegeneration in AD.

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Notes

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ABBREVIATIONS

AD, Alzheimer's disease; $A\beta$, amyloid- β protein; aCSF, artificial cerebrospinal fluid; BSA, bovine serum albumin; GuHCl, guanidine hydrochloride; HFIP, hexafluoroisopropanol; SEC, size-exclusion chromatography; ThT, thioflavin T; Tris, tris(hydroxymethyl)aminomethane.

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